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Photoluminescent properties of oxidized stochiometric and carbon-rich amorphous $Si_{1-x}C_x$:H films

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Abstract. Near-stochiometric and carbon-rich $a-Si_{1-x}C_x$:H thin films were deposited using the magnetron sputtering of Si target in Ar/CH_4 gas mixture. As-deposited nearstochimetric (x = 0.5) sample showed weak blue photoluminescence (PL), while PL of as-deposited carbon-rich (x = 0.7) sample was 20 times stronger and white in color. The films were annealed in pure argon, wet argon, and dry oxygen at 450 °C for 30 min. The intensity of PL in a-Si_{1-x}C_x:H layers were enhanced by the factor from 2 to 12 after annealing in dependence on the annealing atmosphere. The strongest oxidation and strongest light emission were observed in carbon-rich series (x = 0.7) after annealing in oxidizing atmosphere. Structural properties of the films were characterized by infra-red absorption spectroscopy, ellipsometry and electron paramagnetic resonance. The effect of carbon enrichment of a-Si_{1-x}C_x:H films and annealing atmosphere on the evolution of photoluminescence and local interatomic bonding structure in annealed material were studied and analyzed. It has been found that main effects of thermal treatments is strong enhancement of photoluminescence accompanied by formation of Si:C-H_n and Si-O_xC_v bonding. The strongest oxidation effect as well as strongest photoluminescence were observed in carbon-rich a-SiC:H films.

Keywords: a-Si_{1-x}C_x:H, photoluminescence, ellipsometry and electron paramagnetic resonance, infra-red absorption spectroscopy.

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1. Introduction

Development of new white light emitting materials synthesized using the processes compatible with common silicon based technology is of great interest especially in the field of flat displays and indication panels. Recently, it has been demonstrated that carbon incorporated silicon oxide nanostructured materials exhibit a strong white photoluminescence (PL). Intriguing is that white PL was demonstrated by SiOC materials fabricated by quite different methods, namely: SiOC thin films deposited by thermal chemical vapor deposition (TCVD) [1], atmospheric pressure microplasma jet deposition [2], magnetron sputtering deposition [3], C^+ implantation into SiO₂ [4-6], chemical vapor deposition technique [7], carbonization/oxidation of porous silicon [8, 9], sol–gel method [10].

The origin of white light emission in these materials is still unclear but common features of the composition and PL properties indicate a common lightemission mechanism. It is obvious that photoluminescence is associated with incorporation of carbon in silicon oxide matrix but more detailed structural identification is under debate. In this paper, we analyze correlations of PL and structural properties of thin films fabricated by combination of magnetron sputtering deposition and thermal annealing/oxidation of hydrogenated amorphous silicon-carbon alloy films (a-Si_{1-x}C_x:H).

2. Experimental

a-Si_{1-x}C_x:H films were deposited on *p*-type 40 Ohm·cm Si(100) and glass substrates by reactive dc-magnetron sputtering of a silicon target in Ar/CH₄ gas mixture by analogy with [11] and [12]. Using this sputtering configuration, the films with different carbon content can be deposited by varying the substrate-target distance. The larger is the distance, the smaller incorporation of deposited silicon atoms is and, correspondingly, the larger relative contribution of carbon incorporated with hydrocarbon radicals from operation gas. Two samples with x = 0.5 (320 nm thick) and x = 0.7 (780 nm thick) were deposited at the substrate temperature 200 °C. Composition of the films was determined using Augerelectron spectroscopy with SiC(6H) crystal as a reference sample. The volume density of the layers measured by weighing the substrate before and after deposition was 2.2±0.1 g/cm³ (x = 0.5) and 1.6±0.1 g/cm³ (x = 0.7). After the deposition, the samples were annealed at atmospheric pressure in the flow of dry Ar, wet-Ar and dry O₂ flow at 450 °C for 30 min.

3. Results and discussion

PL of as-deposited and annealed $a-Si_{1-x}C_x$:H samples was measured at room temperature using excitation by 351 nm line of Ar^+ laser. The refractive index was calculated using spectral ellipsometry measurements. The local bonding structure in the films was studied by Fourier-Transform Infra-Red spectroscopy (FTIR). Paramagnetic defects were analyzed using electron paramagnetic resonance (EPR) in the X-band with 100-kHz modulation of magnetic field.

Weak PL band with the maximum intensity near 420 nm was detected in the as-deposited $a-Si_{0.5}C_{0.5}$:H film (Fig. 1a, spectrum 1). Annealing at 450 °C for 30 min in dry argon results in appearance of the emission band with the maximum intensity near 510 nm, while the blue component seems to be unchanged (Fig. 1a, spectrum 2). Oxidation by wet argon and oxygen resulted in further development of the "green" band, but the integrated intensity of PL in the sample annealed in wet argon was 5 times higher (Fig. 1c). The narrow features at 700 and 727 nm available in all the spectra are laser satellites. The amplitude of these laser lines can be used for rough estimation of the relative PL intensity in different samples.

As-deposited carbon-rich a-Si_{0.3}C_{0.7}:H film emitted broad PL band white in color with strongly modulated spectrum (Fig. 1b, spectrum 1). We unambiguously attribute modulating features of the spectra to interference effects because: (1) these features are shifted with changing of the angle between sample plane and optical slip of monochromator; (2) these features disappeared after mechanical roughening of the film surface by diamond past (grain size of 20 nm). The maximum intensity of PL in the as-deposited sample was estimated to be within the range 500...550 nm. The PL intensity increased strongly after annealing in dry Ar, while the spectral shape of PL was almost unchanged (Fig. 1b, spectrum 2). Annealing in wet argon and dry



Fig. 1. PL spectra of as-deposited (spectrum *I*) and annealed $(2 - \text{dry Ar}, 3 - \text{wet Ar}, 4 - \text{dry O}_2)$ a-Si_{0.5}C_{0.5}:H (*a*) and a-Si_{0.3}C_{0.7}:H (*b*) films; integrated PL intensities of as-deposited and annealed a-Si_{0.3}C_{0.7}:H (*c*) and a-Si_{0.5}C_{0.5}:H (*d*) films.

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Fig. 2. (*a*) – FTIR spectra of a-Si_{0.5}C_{0.5}:H: (*1*) – as-deposited, and annealed in dry Ar (2), wet Ar (3) and dry O₂ (4); (*b*) – scaled-up spectral range 1220...1300 cm⁻¹ (selected by dashed quadrangle in (*a*)).

oxygen resulted in further increase of PL intensity (Fig. 1d) and the obvious blue shift of the intensity distribution caused by oxygen treatment (Fig. 1b, spectra 3 and 4).

Transmission FTIR spectra of as-deposited and annealed $a-Si_{1-x}C_x$:H films are presented in Fig. 2. The spectrum of as-deposited sample is composed by strong absorption bands at 780 cm⁻¹ (Si–C stretching) and 1010cm⁻¹ (deformation C–H_n and/or stretching Si–O– Si), and weaker absorption bands at 2100 cm⁻¹ (Si–H_n stretching) and 2690...3000 cm⁻¹ (C(sp³)–H_n stretching). A traces of absorption were found at about 1260 cm⁻¹ (deformation modes of Si:C–H₃ bonds) and within range 3000...3500 cm⁻¹ (O–H stretching). The weak narrow spectral features at 608 and 1100 cm⁻¹ are due to imperfect subtraction of the absorption of Si substrate (oxygen contaminations in Si wafer). FTIR absorption peaks and corresponding vibration modes assigned in a-Si_xC_vO_z:H films are presented in Table.

The absorption band at 1010 cm^{-1} in $a-\text{Si}_{1-x}\text{C}_x$:H films is commonly ascribed to rocking/waging vibration modes in CH₂ radicals bonded to silicon atoms $-\text{Si-CH}_2$ -Si [23-27]. But it is well known that the main absorption band in silicon sub-oxide SiO_x (x < 2) is also located at $1000...1400 \text{ cm}^{-1}$ depending on x [20]. Residual oxygen contamination in our films was estimated using Auger electron spectroscopy to be up to 7...10 at.%, so that the contribution of Si-O vibrations should be taken into account. Unfortunately, it is not possible to separate correctly hydrogen- and oxygen-related absorption bands, but, basing on our previous experience [12], we assume that hydrogen-related contribution dominates in as-deposited samples. From Fig. 2a, one can see that the relative intensity of this

band is increased after thermal treatments regardless of annealing atmosphere (spectra 2, 3, 4). Annealing in dry and wet Ar resulted in development of the Si:C–H₃-related band at 1260 cm⁻¹ (Fig. 2b, spectra 2, 3), while the intensity of Si–H_n (2100 cm⁻¹) and C–H_n (2690...3000 cm⁻¹) decreased.

Table. Identification of IR absorption bands in a-SiC:H/ a-SiOC:H thin films.

Absorption	Attribution	References	
band (cm ⁻¹)		Theory	Experiment
450	Si-O-Si, rock.	[13]	[13], [20-22]
780	Si–C, str.	[14-17]	[20], [23-28]
780-800	C–H ₃ , bend.	[16-18]	[27], [29]
800	Si–O–Si, str.	[13]	[13], [20-22]
940	Si–OH, str.		[30-31]
1010	CH ₂ rock/wag	[16-18]	[23-27]
	Si–O in SiO _x (x<2), str.		[20], [32]
1060-1200	Si–O–Si, str.	[13]	[13], [20-22]
1260	Si:CH ₃ , bend.	[16-18]	[23-28], [33], [34]
1700	C=O, str.		[33]
2100	Si–H _n , str.	[16-19]	[19], [23-28]
2700-3000	$C(sp^3)-H_n$, str.	[16-18]	[23-28], [33]
3000-3700	O–H, str.		[21], [35],[36]



Fig. 3. (*a*) – FTIR spectra of a-Si_{0.3}C_{0.7}:H: (*1*) – as-deposited, annealed in dry Ar (2), wet Ar (3) and dry O₂ (4); (*b*) – scaled-up spectral range 1220...1300cm⁻¹ (selected by dashed quadrangle in (*a*)).

FTIR spectra of $a-Si_{0.3}C_{0.7}$:H films are shown in Fig. 3a. The main absorption bands were found to be centered near 790 and 1010 cm⁻¹, correspondingly. The relative intensity of 1010 cm⁻¹ band as well as Si:C-H₃ bending (1256 cm⁻¹) and C-H_n stretching bands (2800... 3000 cm⁻¹) is much stronger in the carbon-rich sample than that in $a-Si_{0.5}C_{0.5}$:H indicating larger contribution of carbon-hydrogen bonds. The OH-related absorption band at 3500 cm⁻¹ is also much stronger.

The intensity of 1010 and 1256 cm^{-1} bands was increased after annealing in dry and wet argon (Figs. 3a and 3b). Comparing Fig. 3a and Fig. 3b, one can find the direct correlation for the intensities of these bands. Absorption intensity of Si–H_n and C–H_n stretching vibration bands decreased after annealing in dry and wet Ar, and completely vanished after thermal treatment in oxygen. Obvious absorption at about 1700 cm⁻¹ due to C=O stretching (sym./asym.) vibrations [33] was observed after annealing in dry and wet argon (Fig. 3a, spectra 3 and 4).

Annealing of a-Si_{0.3}C_{0.7}:H film in oxygen resulted in strong oxidation of the material. Several oxygen related bands are now present in the spectrum (Fig. 3a, spectrum 4): 450 cm⁻¹ (rocking Si–O–Si), 802 cm⁻¹ (symmetrical stretching Si-O-Si), 940 cm⁻¹ (Si-OH stretching), 1063 cm⁻¹ (in-phase asymmetrical stretching Si–O–Si) with characteristic shoulder at 1160 cm⁻¹ (outof-phase asymmetrical stretching), and 3000...3700 cm⁻¹ (O-H stretching). It is logical to assign the weak band at 1279 cm^{-1} (Fig. 3b, spectrum 4) to Si-CH₃ band strongly shifted to the high frequency side due to increase of bonding polarity caused by oxygen incorporation [37]. The weak but well detectable double band within the range 1540...1780 cm⁻¹ (peaks near 1600 and 1700 cm⁻¹) is attributed to C=C and C=O bonds, respectively [33].

Electron paramagnetic resonance measurements (EPR) were performed at room temperature in the X band employing 100-kHz field modulation. The single EPR line with g-factor 2.0026 (±0.0002) was detected in the as-deposited samples. No change of g-factor was observed after thermal treatments. The concentration of paramagnetic centers N_s in the as-deposited a-Si_{0.5}C_{0.5}:H and a-Si_{0.3}C_{0.7}:H films was calculated to be about $2 \cdot 10^{20}$ and $5 \cdot 10^{19}$ cm⁻³, respectively. Variation of the spin concentration in the films after annealing was similar, i.e. gradual decrease after annealing in sequence (dry Ar) – (wet Ar) – (dry O₂).



Fig. 4. Concentration of paramagnetic centers N_s in asdeposited and annealed a-Si_{0.5}C_{0.5}:H and a-Si_{0.3}C_{0.7}:H series. Inset represents EPR spectra Si_{0.3}C_{0.7}:H series.



Fig. 5. Refractive index of as-deposited and annealed $a-Si_{0.5}C_{0.5}$:H and $a-Si_{0.3}C_{0.7}$:H series.

Calculation of the refractive index was performed using ellipsometry measurement in frame of the threelayer model (air/a-Si_{1-x} C_x :H/c-Si). The calculated refractive indexes of all the samples at the 500-nm wavelength are presented in Fig. 5. The refractive index of the as-deposited near-stochiometric a-Si_{0.5}C_{0.5}:H film is about 2.0, which is considerably lower than that of crystalline SiC ($n_{SiC} = 2.5$) due to the lower density. Increase of the carbon content in the as-deposited film up to x = 0.7 resulted in decrease of the refractive index down to 1.8. This is quite consistent with the density of films: about 2.2 g cm^{-3} for near-stochiometric films and about 1.6 $g \cdot cm^{-3}$ for carbon-rich samples, which is significantly lower than 3.2 g cm⁻³ for crystalline SiC. No change in the refractive index was observed after annealing in dry argon, while decrease of the refractive index down to 1.9 was observed in $a-Si_{0.5}C_{0.5}$:H series after oxidation treatments. More significant decrease down to 1.7 and 1.4 was observed in carbon-rich series after thermal treatments in water vapor and oxygen, respectively.

The reduced density of carbon-rich a-SiC:H films is mostly associated with enhanced porosity of the carbon-rich structure [16]. Enhancement of nanoporosity due to development of termination of structural network by $-CH_n$ radicals during film growth is quite similar to that of a-SiOC:H films [38].

It was widely reported that decrease of the carbon content x in a-Si_{1-x}C_x:H films in a broad range results in increase of the refractive index from 1.8...2.1 in carbon-rich to 3.4...3.8 in silicon-rich species [24, 39-40]. This variation is obviously related with variation of relative contribution of C–C, Si–C and Si–Si bonds as well as nanoporosity of the structure.

It is worth noting that FTIR spectra indicate strong oxidation of carbon-rich series by dry oxygen in contrast to much weaker oxidation by water vapor. It is well known that oxidation rate of crystalline silicon carbide by water vapor is considerably higher than that by oxygen [41]. Detailed comparative study of oxidation effects in a-SiC and a-SiC:H caused by thermal annealing in water vapor and oxygen ambient was presented in [42]. The authors suggested that strong oxidation efficiency of oxygen in a-SiC:H is related with its high reactivity against carbon precipitates (i.e. C–C bonds, hydrocarbon radicals, carbon clusters) natively available in hydrogenated amorphous silicon-carbon alloy films.

In Fig. 2, we see that PL efficiency in the films is related to (1) composition of as-deposited layer, (2) thermal effects (annealing in inert atmosphere, e.g. dry argon) and (3) oxidation effects. From the "classical" viewpoint, PL efficiency in amorphous VIgroup semiconductors (a-Si:H, a-C:H, a-SiC:H) is mainly determined by competition of radiative recombination from the conduction band tail to the valence band one (tail-to-tail transition) and nonradiative recombination path through mid-gap paramagnetic defect states [25]. The efficiency of radiative recombination in this model is reversely proportional to the concentration of non-radiative recombination centers associated mainly with paramagnetic dangling bonds.

The concentration of paramagnetic defects of 10^{20} cm^{-3} observed in as-deposited films is too high for effective radiative recombination through the tail-to-tail mechanism, so that blue light emission in as-deposited $a-Si_{0.5}C_{0.5}$:H films may be hardly assigned to recombination in the amorphous SiC matrix. Moreover, the optical band gap value derived from optical absorption measurements of this film on a glass substrate is close to 2.5 eV that is significantly lower than the energy of PL (2.9 eV). We suggest that blue emission is associated with SiO_x structural fragments available in asdeposited films due to residual oxygen contamination. The photon energy of PL band with the maximum intensity near 510 nm (2.4 eV) that arose after annealing is close to the optical band gap value, so that this band may be assigned to tail-to-tail recombination in the amorphous SiC network.

Comparing evolution of the PL intensity (Fig. 1) and concentration of paramagnetic defects (Fig. 4) in a-Si_{0.5}C_{0.5}:H series, one can find no correlation. Moreover, comparing the PL intensity and concentration of paramagnetic centers in the a-Si_{0.5}C_{0.5}:H sample annealed in dry argon (N_s is about $6 \cdot 10^{19}$ cm⁻³) and in the asdeposited a-Si_{0.3}H_{0.7}:H sample (N_s is about $4 \cdot 10^{19}$ cm⁻³), one can see that the PL intensity differs by 10 times, while the concentration of paramagnetic centers remains almost the same. It is obvious from these observations that the concentration of paramagnetic defects is not the only parameter determining the PL efficiency in a-Si_{1-x} C_x :H. In the previous report [12], it was suggested that the PL efficiency in $a-Si_{1-x}C_x$:H films can be increased by enhancement of spatial localization of photo-excited electron-hole pairs caused by increasing the concentration of C-H bonds and corresponding increase of spatial fluctuations of the interatomic potential. In frame of this hypothesis, one can reasonably suggest that introducing Si-O bonds would also increase

fluctuations of this potential and enhances localization of photo-excited electron-hole pairs in the SiC network.

From Fig. 2b, one can see that the absorption band at 1263 cm⁻¹ is the strongest one after annealing in wet argon. Taking into account obvious incorporation of oxygen due to interaction with water vapor, one may expect the strongest localization in this sample that is in good agreement with the highest PL intensity. After annealing in oxygen, the 1263-cm⁻¹ band is very weak, so that localization is mainly determined by oxygen incorporation, and PL intensity is lower.

The interferential feature in PL spectra of asdeposited a-Si_{0.3}C_{0.7}:H does not allow correct determination of the maximum intensity in the PL band, but rough estimation is about 550 nm (2.3 eV) for the asdeposited sample, while the optical band gap of this sample is about 2.7 eV. Combination of the effects of passivation of paramagnetic centers and enhancement of localization by C-H and Si-O bonds well explains increase of the PL intensity after annealing in dry and wet argon, but it is not the case after oxygen treatment, when almost full oxidation is observed (Fig. 3, spectrum 4). The PL spectrum is strongly changed and obviously shifted to blue region after annealing in dry oxygen (Fig. 1b, spectrum 4), so that one may reasonably expect that light emission happens not in amorphous SiC matrix. FTIR spectra identify the structure of oxidized film as silicon oxide matrix with some incorporation of carbon precipitates, consequently PL can be attributed presumably to both of these structural domains.

Photoluminescence of undoped nanostrucutred silica are commonly considered in terms of optically active point defects (reviewed in [45]). But photon emission of these defects in the visible spectral range is associated with the triplet transition $(T_1 \rightarrow S_0)$, i.e. "phosphorescence", that is the natively slow process with a characteristic decay time of micro- and milliseconds, or even seconds. However, time-resolved measurements showed that the PL decay time in oxidized carbon-rich series is of about 10 nanoseconds.

Direct correlation of the broad band PL intensity and concentration of Si–O–C bonding configuration in a-SiOC thin films was demonstrated in [1]. The concentration of Si–O–C bonds was evaluated using the IR absorption band at 440 cm^{-1} . In [2], the same authors developed their hypothesis and suggested that the oxygen vacancy in Si–O–C bridges also contributes to broad band light emission.

Carbon nanoclusters are also to be considered as a possible source of visible light emission. Recently, it was demonstrated that nanosized carbon particles in the form of 3D nanodots, 2D graphene nanodots and 2D-3D graphene oxide (GO) nanodots synthesized under proper conditions exhibit an effective visible photoluminescence with the decay time of the order of several nanoseconds [46-48].

4. Conclusions

More detailed studies are necessary to identify the light emission mechanism reliably. In summary, thermal and oxidation effects on photoluminescence in nearstochiometric and carbon-rich a-SiC:H films deposited using dc-magnetron sputtering have been studied. It has been found that the main effect of thermal treatment is strong enhancement of photoluminescence accompanied by formation of Si:C–H_n and Si–O_xC_y bonding. The strongest oxidation effects as well as strongest white photoluminescence were observed in carbon-rich a-SiC:H films.

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