Semiconductor Physics, Quantum Electronics & Optoelectronics. 2002. V. 5, N 1. P. 76-77.

PACS: 84.37; 84.32.T; 84.60.V

Study of supercapacitors with a double electrical layer based on activated carbon materials

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Abstract. It is proposed ecologically pure technology to obtain activated carbon. On the base of this carbon the supercapacitors were manufactured. Their characteristics were determined and compare with analogs obtained using ecologically dangerous materials.

Keywords: activated carbon; supercapacitor; ecologically pure technology.

Paper received 17.10.01; revised manuscript received 24.12.01; accepted for publication 05.03.02.

1. Introduction

Technological process for obtaining activated carbon (AC) with highly-developed surface (>1000 m²/g) for manufacturing supercapacitor (SC) with a double electric layer is complex, often ecologically dangerous process [1,2]. Therefore, search of new possibilities to obtain AC that improve fabrication process and eliminate existing defects seems to be urgent. In the present work possibilities of obtaining AC in closed volume were studied when carbonization and activation of initial material take place without inert gas as a carrier of pore generator.

In contrast to traditional technologies, when flow of vapor and inert gas takes place at the same temperature, respectively, outflow this mass plus reaction products take place at another temperature, in our case, mass and heat transport are practically excluded. That allows to control effectively technological regime of obtaining AC.

Thus, the aim of the present work is to obtain activated carbon with developed surface area by ecologically pure technology, to manufacture supercapacitors on its base and to determine their characteristics.

2. Results and discussion

As an initial material for obtaining AC ecologically pure material, fruit pips (cherry, plum, cherry sweet) were used, while in majority technologies ecologically dangerous (especially at heating) organic compounds (formaldehyde, hydrocelluloze, sterol, etc) are used. As a pore generator 5÷20% water solution of KOH was used. Milled initial material was heated in vacuum with residual pressure less than 10² mm Hg and the temperature at which their full carbonization takes place. Then obtained material in mixture with 0÷20% water solution of KOH was heated in closed container up to 840÷880°C and kept up at this temperature during 70+80 min. In both cases heating was linear in time with the rate of 15÷20 °/min and cooling was in the turn-off furnace regime. As a consequence, carbonization of initial material and its activation take place. It is evident that with quantity of water or KOH solution one can easily control the vapor pressure, and in this way influence on process of pore generation and their sizes can be observed. At that temperature regime, the process is easily controlled since only pipe-bend of water vapor with decay products takes place.

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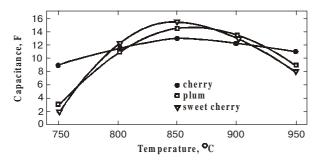


Fig. 1. Dependence of the capacitance of SC on the activation temperature of AC from which electrodes are formed.

Fig. 1 shows experimental dependence of the SC capacitance type in the body with the typesize '2525', formed on the base of obtained material, on temperature of activation. As it is seen from Fig. 1, optimal temperature for activation of preliminary carbonized initial material is $840 \div 880$ °C under other identical conditions. In this case, the internal resistance at mentioned temperatures lies in the range $0.08 \div 0.1 \Omega$. Achieved parameters of SC are not worse than the analogous ones formed on the base of AC from stirolvinilbenzol copolymer[3].

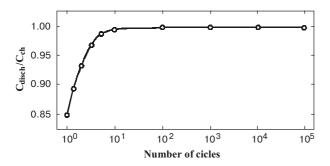


Fig. 2. The Coulomb efficiency of SC from obtained AC in the body typesize '2525'.

Fig. 2 shows the Coulomb efficiency of SC formed on the base of obtained AC. As it is seen from Fig. 2, it is already after some cycles the Coulomb efficiency was stable at the level close to unity and was constant up to investigated 10^5 cycles. This confirms that Faradey reactions in investigated SC are absent what is intrinsic to the ideal polarized electrods. Moreover, it is found that the typical charge-discharge curves at the charge current 0.05 A and at the discharge current 0.02 A are normal. This indicates that investigated capacitors are close to the classic ones.

Dependencies of obtained SC on the discharge current (Fig. 3) were also studied. Small differences in these dependencies as compared to SC formed using AC from stirolvinilbenzole copolymer were found.

Thus, the process of manufacturing AC is simplified as well as more controlled and ecologically secure. At the same time, operational characteristics of SC formed on its base is not worse than the analogous ones obtained using ecologically dangerous materials.

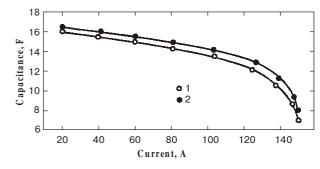


Fig. 3. Dependence of the capacitance of SC at typesize '2525' on the discharge current. 1- SC on the base of obtained AC, 2-SC on the base of AC obtained from stirolvinilbenzole copolymer.

3. Conclusions

We proposed and realized a new technology for obtaining activated carbon. This technology uses ecologically pure materials. All technological process is also ecology secure and occurs in a closed volume without any inert gas. We obtained activated carbon with the developed surface area 900-1000 m²/g and manufactured supercapacitors on its base. Their capacities in the body with the typesize '2525' are about 10÷15 F and the internal resistances lie within the limits 0.06÷0.09 Ω .

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